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# Diploma in Pharmacy ${ }^{\text {st }}$ Year Pharmaceutics 

## Experiment

To prepare 50 g of cold cream.
Aim:
To prepare 50 g of cold cream.

## Reference :

' Dr. Gupta G.D , Dr. Sharma Shailish , Dr. Sharma Neelam'<br>"Practical Manual of Pharmaceutics" Published by Nirali Prakashan, Page no 86 - 90<br>\section*{Requirment :}

Cetyl ester wax, white wax, mineral oil, sodium borate, purified water.

## Theory :

The term cream in pharmacy and medicine is applied to viscous emulsions or semisolid preparations consisting of solutions or dispersions of one or more medicaments in suitable base and intended for application to the skin or mucus membrane. They are applied to the skin for protective, beautifying, therapeutic or prophylactic purposes. Creams may contain suitable antimicrobial or preservatives unless the medicaments or basis have sufficient intrinsic bactericidal and fungicidal activity.

Cold cream is a water-in-oil emulsion. It is useful as an emollient, cleansing cream and ointment base. It resembles rose water ointment differing only in that mineral oil is used in place of almond oil and omitting the fragrance. This change produces an ointment base which is not subjected to rancidity like one containing a vegetable oil.

Category: Emollient and cleansing cream.
Uses: It is used as a cleansing cream.

Formulation Table for Cold Cream

| Ingredient | Quantity required for <br> $\mathbf{1 0 0 \mathbf { g }}$ | Quantity required for <br> $\mathbf{5 0 g}$ |
| :---: | :---: | :---: |
| Cetyl ester wax | 12.5 g | 6.25 g |
| White Wax | 12.0 g | 6.0 g |
| Mineral Oil | 56.0 ml | 28.0 ml |
| Sodium Borate | 0.5 g | 0.25 g |
| Purified Water | 19.0 ml | 9.50 ml |

## Calculation

Cetyl ester wax required to prepare 100 g of cream $=12.5 \mathrm{~g}$
So, for preparation of, 50 g of cream,
Cetyl ester wax required $=12.5 / 100 * 50=6.25 \mathrm{~g}$
White wax required to prepare 100 g of cream $=12.0 \mathrm{~g}$
So, for preparation of 50 g of cream,
White wax required $=12.0 / 100^{*} 50=6.0 \mathrm{~g}$
Mineral oil required to prepare 100 g of cream $=56.0 \mathrm{ml}$
So, for preparation of 50 g of cream,
Mineral oil required $=56.0 / 100$ * $50=23.0 \mathrm{ml}$
Sodium borate required to prepare 100 g of cream $=0.5 \mathrm{~g}$
So, for preparation of 50 g of e of cream,
Sodium borate required $=0.5 / 100{ }^{*} 50=0.25 \mathrm{~g}$

Procedure : Reduce size of cetyl ester wax and white wax in small pieces. Melt them on a steam bath with the mineral oil and continue heating until the temperature of the mixture reaches $70^{\circ} \mathrm{C}(\mathrm{A})$. Dissolve the sodium borate in purified water and heat at $70^{\circ} \mathrm{C}$ in separate flask (B). Gradually add aqueous solution (B) to the mixture of oily phase (A) with continuous and rapid stirring until it has congealed.

Storage : Preserve in a well-closed container

## Label :



## Result :

It was found that 50 g of Cold Cream was successfully prepared and submitted.

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